

# Liposomal Formulation of Trilaciclib Injection, With Hydrogenated Palm Oil

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#### ABSTRACT

Trilaciclib dihydrochloride, is a kinase inhibitor, chemically; he chemical name for Trilaciclib is 2'-{[5-(4-methylpiperazin-1-yl)pyridin-2-yl]amino}-7',8'dihydro-6'H-spiro[cyclohexane-1,9'-

pyrazino[1',2':1,5]pyrrolo[2,3-d]pyrimidin]-6'-one. The inventors of the present invention have surprisingly found that it is possible to prepare a stable lipid Nano composition of Trilaciclib. Liposomal formulations are less toxic than drugs alone and have betterpharmacological parameters. Although they seem to be the first choice for drugdelivery systems for various diseases.

The usage of natural products in pharmaceuticals has steadily seen improvements over the last decade, and this study focuses on the utilization of palm oil in formulating liposomal Trilaciclib. The liposomal form of Trilaciclib generally minimizes toxicity and enhances target delivery actions. Taking into account the antiproliferative and antioxidant properties of palm oil, the aim of this study is to design and characterize a new liposomal Trilaciclib by replacing phosphatidylcholine with 5% and 10% palm oil content. Liposomes were formed using the freeze thaw method, and Trilaciclib was loaded through pH gradient technique and characterized through in vitro and ex vivo terms. Based on TEM images, large lamellar vesicles (LUV) were formed, with sizes of 438 and 453 nm, having polydispersity index of 0.21  $\pm$  0.8 and  $0.22 \pm 1.3$  and zeta potentials of about -31 and -32 mV, respectively. In both formulations, the entrapment efficiency was about 99%, and whole Trilaciclib was released through 96 hours in PBS (pH = 7.4) at 37°C. Comparing cytotoxicity and cellular uptake of LUV with on MCF7 and MDA-MBA 231 breast cancer cell lines indicated suitable uptake and lower IC50 of the prepared liposomes.

**Keywords:**Trilaciclib, Liposomal,Hydrogenated palm oil:polyethylene glycol, L-alphaphosphatidylcholine. Methanol, chloroform, Sodium hydroxide, potassium dihydrogen phosphate,Malvern Zetasizer, Zeta Potential, Polydispersity Index Trilaciclib dihydrochloride is commercially available COSELA (trilaciclib) for injection, which is a lyophilized formulation available as 300 mg / vial and (trilaciclib) for injection is a yellow lyophilized cake supplied in a single-dose vial. Each vial contains one 300 mg strength single-dose vial. The current investigation was designed to alternative stable liposomal formulations.

**Drug Substance:** Trilaciclib dihydrochloride is a water-soluble yellow solid.

**Chemical Properties:** COSELA for injection contains trilaciclib dihydrochloride, a kinase inhibitor. The chemical name for trilaciclib is 2'-{[5-(4-methylpiperazin-1-yl)pyridin-2-yl]amino}-7',8'-dihydro-6'H-spiro[cyclohexane-1,9'-

pyrazino[1',2':1,5]pyrrolo[2,3-d]pyrimidin]-6'-one. Trilaciclib has the following structure:



**Mechanism of Action:**Trilaciclib is a transient inhibitor of CDK 4 and 6. Hematopoietic stem and progenitor cells (HSPCs) in the bone marrow give rise to circulating neutrophils, RBCs, and platelets. HSPC proliferation is dependent on CDK4/6 activity.

**Materials:**Trilaciclib, Hydrogenated palm oil (palm oil), cholesterol (CH), L-alphaphosphatidylcholine (PC), polyethylene glycol (PEG), methanol, and chloroform were purchased from Sigma-Aldrich. Sodium hydroxide and potassium dihydrogen phosphate were purchased from Merck.

**Methods**: Liposomes were prepared using the freeze-thaw method and pH gradient technique, carried out in order to maximize the loading of Trilaciclib, within liposome. Two formulations



were designed; both consisted of 45 mg CH and 5 mg PEG with different percentages of PC and palm oil. The first formula (Fa) contains 5% palm oil and 45% of PC, while the second formula (Fb) contains 10% palm oil and 40% PC in their respective formulations. Then, all of the lipid components and PEG were dissolved in a chloroform: methanol mixture of (2:1, v/v) in a round-bottom flask. The solvent was removed under vacuum using a rotary evaporator (Rotavapour R-124, BÜCHI) at 40°C and 50 rpm. After a thin lipid film was formed in the interior of the flask, the system was purged with nitrogen to remove organic solvent entirely. The lipid film layer was hydrated with 10 mL Citrate buffered solution (pH = 4) and then sonicated for 30 minutes in a bath type sonicator (Sonicor). The freeze-thaw cycle was carried out five times via freezing under -80°C and then heated mixture in water bath at 65°C with the intention of decreasing the size, further entrapping the acidic buffer inside the liposome. Bicarbonate buffer (pH = 0.5) was added dropwise to the mixture (for the reason of adjusting outer liposomes space into a physiological pH) until its pH reaches 7. Afterwards, 10 mL of Trilaciclib medium in distilled water (2000 µg/mL) was added to the mixture and shaken at room temperature for 30 minutes at 60 rmp.

Liposomes are specialized delivery vehicles that serve multiple roles in enhancing the capabilities of active pharmaceutical ingredients (APIs). These lipid bilayers form in the shape of hollow spheres, encapsulating cargo of interest within an aqueous interior or lipid bilayer.



Figure 1.0: Design of Liposomes

Liposomes are a novel drug delivery system (NDDS), they are vesicular structures consisting of bilalyers which form spontaneously when phospholipids are dispersed in water. They are microscopic vesicles in which an aqueous volume is entirely enclosed by a membrane composed of lipid bilayers.

Liposomes are microscopic fat-soluble vesicles derived from lipids. In liposomal delivery, liposomes are used to encapsulate and transport active ingredients in drugs and nutritional supplements to locations in the body where they are most efficiently absorbed.

A liposome is a spherical vesicle having at least one lipid bilayer. The liposome can be used as a drug delivery vehicle for administration of nutrients and pharmaceutical drugs, such as lipid nanoparticles in mRNA vaccines, and DNA vaccines.

Since palm oil has antiproliferative and antioxidant properties due to presence of components such as carotenes, tocopherol, tocotrienols, terpenoids, and flavonoids, it is viable for use in pharmaceutical products, on top of its nutritional advantages. In addition, its antioxidants help resist rancidity and improve the stability of palm oil.

Considering the anticancer properties of palm oil and great advantages of liposome, the aim of this study was to prepare liposomal Trilaciclib by applying palm oil fractions.

## MATERIALS AND METHODS

Trilaciclib, Hydrogenated palm oil(palm oil), cholesterol (CH), L-alpha-phosphatidylcholine (PC), polyethylene glycol (PEG), methanol, and chloroform were purchased from Sigma-Aldrich. Sodium hydroxide and potassium dihydrogen phosphate were purchased from Merck.

Liposomes were prepared using the freeze-thaw method and pH gradient technique, carried out in order to maximize the loading of Trilaciclib, within liposome . Two formulations were designed; both consisted of 45 mg CH and 5 mg PEG with different percentages of PC and palm oil. The first formula (Fa) contains 5% palm oil and 45% of PC, while the second formula (Fb) contains 10% palm oil and 40% PC in their respective formulations. Then, all of the lipid components and PEG were dissolved in a chloroform : methanol mixture of (2:1, v/v) in a round-bottom flask. The solvent was removed under vacuum using a rotary evaporator (Rotavapour R-124, BÜCHI) at 40°C and 50 rpm. After a thin lipid film was formed in the interior of the flask, the system was purged with nitrogen to remove organic solvent entirely. The lipid film layer was hydrated with 10 mL Citrate buffered solution (pH = 4) and then sonicated for 30 minutes in a bath type sonicator (Sonicor). The freeze-thaw



cycle was carried out five times via freezing under  $-80^{\circ}$ C and then heated mixture in water bath at  $65^{\circ}$ C with the intention of decreasing the size, further entrapping the acidic buffer inside the liposome. Bicarbonate buffer (pH = 0.5) was added dropwise to the mixture (for the reason of adjusting outer liposomes space into a physiological pH) until its pH reaches 7. Afterwards, 10 mL of Trilaciclib medium in distilled water (2000 µg/mL) was added to the mixture and shaken at room temperature for 30 minutes at 60 rmp.

#### FORMATION AND MORPHOLOGY

The formation of liposomes was observed with a transmission electron microscope (TEM). Samples were prepared by applying a drop of the mixture to a carbon-coated copper grid and left for a minute to allow some of the particles to adhere onto the carbon substrate. After removing the excess dispersion with a piece of filter paper, a drop of 1% phosphotungstic acid solution was applied for one minute and then left to be air-dried. The samples were viewed with a TEM.

# Particle Size Distribution, Polydispersity Index (PDI), and Zeta Potential (ZP) Measurement

To evaluate the size distribution, PDI, and value ZP of each sample, 50 mg of liposome was weighted and dispersed in 20 mL distilled water and then those parameters were measured by the zetasizer (Zetasizer Nanoseries, Malbern Instrument). This test was repeated thrice.

#### Liposome Particle size distribution (PSD) Introduction:

Dynamic Light Scattering (also known as Photon Correlation Spectroscopy or Quasi-Elastic Light Scattering) is a technique for measuring the size of particles typically in the sub-micron region.

Dynamic Light Scattering (DLS) measures Brownian motion and relates this to the size (Hydrodynamic diameter) of the particles. The relationship between the size of a particle and its speed due to Brownian motion is defined by Stokes-Einstein equation.

Hydrodynamic Diameter: The size of a particle is calculated from the translational diffusion coefficient by using the Stokes-Einstein equation; D (H) =  $kT / 3\pi\eta D$ 

D (H) = hydrodynamic diameter, D = translational diffusion coefficient

 $k = Boltzmann's constant, T = absolute temperature, <math>\eta = viscosity$ 

It does this by illuminating the particles with a laser and analysing the intensity fluctuations in the scattered light. Brownian motion is the random movement of particles due to the bombardment by the solvent molecules that surround them.

#### Working Principle of Malvern Zeta-Sizer

A typical dynamic light scattering system comprises of six main components. Firstly, a laser (1) provides a light source to illuminate the sample contained in a cell (2). For dilute concentrations, most of the laser beam passes through the sample, but some is scattered by the particles within the sample at all angles.

A detector (3) is used to measure the scattered light. The detector position will be at either  $173^{\circ}$  (in case of Malvern Zetasizer Nano ZS) or  $90^{\circ}$  (in case of Malvern Zetasizer Nano ZS90). The intensity of scattered light must be within a specific range for the detector to successfully measure it. If too much light is detected, then the detector will become saturated.

To overcome this, an attenuator (4) is used to reduce the intensity of the laser source and hence reduce the intensity of scattering. The scattering intensity signal from the detector is passed to a digital processing board called a correlator (5).

The correlator compares the scattering intensity at successive time intervals to derive the rate at which the intensity is varying. This correlator information is then passed to a computer (6), where the software will analyze the data and derive size information.

The fundamental particle size distribution obtained from a dynamic light scattering measurement is based upon the intensity of light scattered by the particles being measured. The mean particle size obtained is the intensity based mean size which is called as "Z Average".

The particle size distribution of Trilaciclib liposome injection lies between approx. 30 nm to 300 nm. So the formulation can be analyzed using Zeta-Sizer instrument with detector positioned either at 173° (Nano S & ZS range: 0.6nm to  $6\mu$ m) or 90° (Nano S90 &ZS90 range:2nm to  $3\mu$ m).

#### **Procedure:**

Approx 1mL Sample was put into disposable Polystyrene cuvette and particle size distribution was analysed using Malvern Zetasizer Nano ZS using Particle absorbance 0.010 and Refractive index 1.35.

The following parameters are considered important during the Particle size analysis:

- Average Particle Size Z<sub>avg</sub>)
- Polydispersity Index (PDI)
- D10, D50 and D90 values
- SPAN (D90-D10)/D50







#### **INTRODUCTION:**

Surface charge/zeta potential on liposomes can affect the clearance, tissue distribution, and cellular uptake.

The magnitude of the zeta potential gives an indication of the potential stability of the colloidal system. If all the particles in suspension have a large negative or positive zeta potential then they will tend to repel each other and there is no tendency to flocculate. If the particles have low zeta potential values then there is no force to prevent the particles coming together and flocculating.

The general dividing line between stable and unstable suspensions is generally taken at either +30mV or-30mV. Particles with zeta potentials more positive than +30mV or more negative than -30mV are normally considered stable.

The Zeta potential analyser calculates the zeta potential by determining the Electrophoretic Mobility and then applying the Henry equation.

The development of a net charge at the particle surface affects the distribution of ions in the surrounding interfacial region, resulting in an increased concentration of counter ions (ions of opposite charge to that of the particle) close to the surface. Thus an electrical double layer exists around each particle.

The liquid layer surrounding the particle exists as two parts; an inner region, called the Stern layer, where the ions are strongly bound and an outer, diffuse, region where they are less firmly attached. Within the diffuse layer there is a notional boundary inside which the ions and particles form a stable entity. When a particle moves (e.g. due to gravity), ions within the boundary move with it, but any ions beyond the boundary do not travel with the particle. This boundary is called the surface of hydrodynamic shear or slipping plane.

The potential that exists at this boundary is known as the Zeta potential.





Concept of Zeta potential and electric double layer

Two values are generally used as approximations for the f(Ka) determination either 1.5 or 1.0. Electrophoretic determinations of zeta potential are most commonly made in aqueous media and moderate electrolyte concentration. f(Ka) in this case is 1.5, and is referred to as the Smoluchowski approximation.For small particles in low dielectric constant media f(Ka) becomes 1.0 and allows an equally simple calculation. This is referred to as the Huckel approximation. Nonaqueous measurements generally use the Huckel approximation.

## Liposome Morphology:

Liposome morphology and degree of lamellarity governs the drug loading, drug retention, and the rate of drug release from the liposomes.

## **Construction of Standard Curve**

Dilutions of Trilaciclib were in the range of 400, 200, 100, 50, 25, and 12.5 ng/mL, prepared and detected by HPLC with a fluorescence detector Mixture of Acetonitrile. Heptanesulfonic acid (0.2%, pH 4) by a ratio of 25 : 75 was applied as mobile phase with the flow rate of 1 mL/min. Trilaciclib has an excitation wavelength of 480 nm and an emission wavelength of 560 nm.

# Evaluation of Entrapment Efficiency and In Vitro Release

The mixture was centrifuged (Universal 32) for 70 minutes at 14000 rpm, the supernatant containing free Trilaciclib was obtained, and the absorbance was measured using HPLC [15]. The entrapment efficiency of liposomes was determined by the following formula:where EE is the concentration of entrapped drug (ng/mL), is the initial concentration of drug used in formulating the liposomes (ng/mL), is the concentration of drug in the supernatant (ng/mL), and EE (%) is the percentage of the drug's entrapment.

To estimate the in vitro drug release of liposomal Trilaciclib, a dialysis bag was used. After separating free drug, 100 mg of liposome was weighted and then placed directly into the dialysis bag (Mw12000). The dialysis bag was sealed at both ends and located in a 1000 mL fresh PBS buffer medium (pH 7.4) at 37°C, at 90 rpm under perfect sink conditions. At predetermined time intervals, 1 mL of the medium was sampled for further analysis by HPLC. The concentrations of Trilaciclib throughout the releasing time were calibrated using the calibration equation. The results recorded are the mean value of the three runs carried out for each liposome concentration. The percentage of released Trilaciclib at certain time was plotted using Microsoft Excel and was defined by the following formula:where is the concentration of drug released (ng/mL) at time and is the initial drug concentration (ng/mL).



#### **Cellular Uptake**

To observe the cellular uptake, two breast cancer cell lines, received from Pasture Institute, were utilized separately. MCF-7 cells were cultured in RPMI 1640, and 10% FBS were then seeded in 24-well plates with a density of  $1 \times 105$  cells/well and incubated in 37°C with 5% CO2 for 24 h. 50 µL Trilaciclib liposome (2000 µg/mL) was added into each well and incubated for 24 h, and then the cells were washed thrice with BPS, respectively. Afterward, image analyses of cells were performed with confocal microscopy (IX71, Olympus, Japan), and the same procedure was carried out for MDA-MBA, 231 cells as well .

#### Cytotoxicity Assay

MTT assay was performed to observe the cytotoxic activities of designed liposome, IC50 of formulation assessed in cell culture media and compared with (pegylated liposomal Trilaciclib). The human breast cell lines MCF 7 and MDA-MBA 231 were seeded in 96-well plates with a density of  $7 \times 103$  cells/well, using RPMI 1640 and 10% FBS added and then incubated in  $37^{\circ}$ C with

5% CO2 for 24 h. The cells were then treated with various concentrations of (2000  $\mu$ g/mL), Fa, and Fb (liposome containing 5% and 10% of palm oil loaded with 2000  $\mu$ g/mL Trilaciclib which is in the same concentration of ), respectively, and then incubated for 48 h. Afterwards, the media were removed, and 10  $\mu$ L MTT was added to each well, incubated for a further four hours. Finally, the MTT was removed, and 100  $\mu$ L DMSO was added to each well, and the absorbance was measured with an ELISA .

#### Statistical Analysis

All of the results were remarked with the mean  $\pm$  SD, and the one-way analysis of variance (ANOVA) was employed for statistical analysis of the data,

#### Results

TEM images in Figure <u>1</u> demonstrate the formation of vesicles. Considering the TEM images, one layer liposome with large inside capacity confirms the fine formation and well shape of the LUV in both formulations.



 $\label{eq:Figure 1} Figure \ \underline{1} \ (Fb) \\ TEM \ images \ of \ Trilaciclib \ liposome \ with \ magnification \ 8000x, \ (a) \ (Fa), \ (b) \ (Fb).$ 



# Particle Size Distribution and Zeta Potential Measurement

Particle size determinations are mostly performed to confirm that the desired liposome size range has been obtained during preparation because suitable size of particles is important for their interaction with the biological situation; for instance, through intravenous administration of loaded particles, their ability to pass or leave the vascular capillaries effectively is dependent on the sizing. Referring to Table 1, Fa has a size of 438 nm, while Fb has a size of 453 nm; the nanosize of LUVs would result in advance drug delivery.

The polydispersity index value is a measure of the heterogeneity of particle sizes in a compound. Liposomes with PDI value between 0.1 and 0.25 display more uniformity and physical stability. Further PDI value more than 0.5 indicates the poor uniformity of mixture. Looking at Table 1, the PDI values of liposomes are 0.22 and 0.21 which confirm the uniformity and homogeneity of LUVs in the mixture as well.

Formulation	Mean particle size (nm, ±SD)	Mean zeta potential (mV, ±SD )	Mean polydispersity index (PDI)	Mean entrapment efficiency (%, ±SD)
Fa	$438.74 \pm 1.9$	$-31.1 \pm 2.6$	$0.22 \pm 1.3$	99.98 ±3.18
Fb	453.71 ± 1.1	$-32.2 \pm 4.1$	$0.21 \pm 0.8$	$99.99 \pm 5.22$

Table 1:	Particle size, ze	ta potential	, and entrap	ment efficiency	y of the	liposomes.

The value of zeta potential (ZP) proves the stability of the particulate systems. It is a measurement of the repulsive forces between the particles. Particles having a ZP of less than -30 mV or more than +30 mV are usually regarded as

stable. Considering the ZP values were higher than -30 mV (Table 2), which confirms the acceptable stability of LUVs as well as their uniformity and size homogeneity suspension.

Formulation	IC50 MCF7 (µg/mL,n =3 )	IC50 MDA-MBA 231 (µg/mL, n=3)
Fa	$376.45\pm9.20$	726.40 ± 7.58
Fb	387.22 ±6.93	$755.73\pm 6.81$

# Table 2: IC50 of Fa, Fb after 48 hours of treatment.

#### **Construction of Calibration Curve**

The following equation from the HPLC results was obtained: Y = 21998 X + 8938, where Y is the area under the curve and X is the concentration of Trilaciclib; the regression line of R2 = 0.999 was obtained as well.

# Entrapment Efficiency and In Vitro Drug Release

As seen from Table 1, in both formulations, liposomes contained maximum entrapment efficiency, nearly 100%, using the pH gradient technique.

Figure 2 shows the in vitro release of Trilaciclib during 96 hours where both formulations demonstrate a constant and continuous release profile. Since Fa and Fb liposomes include same ingredient with only difference in amount of PC and palm oil, they also have comparable releasing pattern with small variation. Within the first 6 hours, Fa demonstrated a faster release rate compared to Fb. During 6–24 hours, Fa and Fb liposomes showed almost similar release whereas, after 36 hours, Fa release goes slower than Fb; however no significant difference is observed.

## Cellular Uptake and Cytotoxicity

Figure 3 demonstrates the cellular fluorescence images and cellular uptake of the Trilaciclib after the cells were incubated with liposome for 24 h. As Trilaciclib emits red fluorescence, the presence of Trilaciclib liposome can be clearly observed in MCF-7 and MDA-MBA 231 cells. After incubation of cell lines with Trilaciclib liposomes, they would cross the cell's membrane and the viable cells appear to have a red basis, while the apoptosis cells exhibited brighter reds, respectively





Figure 3(c)

Figure 4 (d)

Figure 3 -Cellular uptake of Trilaciclib liposome: (a) Fa liposome in MCF 7 cells, (b) Fb liposome in MCF 7 cells, (c) Fa liposome in MDA-MBA 231 cells, and (d) Fb liposome in MDA-MBA 231 cells.

Sr.	Tests	Specification	Trilaciclib injection 10 mg/mLStability condition (5±3°C Inverted)					
		specification						
NO.		Stage	Initial	6M	9M	12M		
1	Description	*	*	*	*	*		
2	Assay (%)	Between 90-110	98.50	94.2	94.20	93.41		
	Lipid content (%)							
	palm oil content	Between 80-120	96.90	92.73	95.33	94.14		
3	3 L-alpha- phosphatidylcholine	Between 80-120	95.30	92.31	88.52	90.11		
	Cholesterol	Between 80-120	97.20	93.79	96.25	96.81		
4	Free Drug (%)	NMT 5% of assay	2.50	1.59	2.57	2.14		
5	Entrapped Drug (%)	NLT 92% of assay	98.50	98.77	96.10	97.96		
6	Zeta Potential (mV)	-5mV to -15mV	-9.38	-9.81	-9.93	-9.02		
7	Osmolality (mOsm/kg)	250-400mOsm/kg	324	327	336	327		
8	рН	Between 6.0 to 7.0	6.49	6.4	6.65	6.62		

Table 3: Stability study results of TrilaciclibInjection 10 mg/mL



Sr.	Tests		Specification	Trilaciclib injection 10 mg/mL Stability condition (5+3°C Inverted)								
No.			Stage		Initial		6M		9M		12M	
	Particle Size Distribution											
	Zavg (nm)	Zavg (nm)			91.7		91.2		93.69		92.28	
9	D10 (nm)		45 to 80		64		63.1		65.20		63.90	
	D50 (nm)		80 to 120		95.4		94.97		97.50		96.20	
	D90 (nm)		120 to 190		143		143	147			145	
	Related substance	es (%	)									
	Impurity I		NMT 0.2 %		0.05		0.03		0.04		0.06	
	Impurity II		NMT 0.8 %		0.14		0.86		0.48		0.66	
	Impurity III		NMT 1.0 %		0.03		0.12		0.10		0.11	
10	Impurity IV		NMT 0.2 %		0.01		0.05		0.08		0.05	
	Impurity V		NMT 0.2 %		0.02		0.04		0.08		0.06	
	Any unspecifing impurity	fied	NMT 0.2 %		0.02		0.13		0.07		0.14	
	Total Impurities		NMT 3.0%		0.09		0.17		0.46		0.41	
	Lipid Degradation Products											
	Lyso–PC content (%)	NMT 6.0%		2	2.19		NA		2.62		3.27	
11	Lyso PE- PEG2000 (%)	NMT 10.0% NMT 2.0%				N			4.33		3.52	
	Stearic acid			N	NA				1.1		1.26	
	Palmitic acid	NMT 0.5%							0.1		0.05	
	In-Vitro Release (%) {Dissolution media pH 6.4; Temperature 52±0.5°C}											
	After 0.5 Hrs.	1.5 Hrs - NLT 15% 3.0 Hrs - NLT 35% 7.0 Hrs - NLT 80%		1	4	2 24		_		2		
10	After 1.5 Hrs.			3	2					1	19	
12	After 3.0 Hrs.			5	7	5	54	NA		5	50	
	After 5.0 Hrs.			8	1	78		1		7	70	
	After 7.0 Hrs.	1			8	9	94				8	

# CONCLUSION

In order to take advantage of the therapeutic effects of palm oil, liposomal Trilaciclib formulations were prepared by replacing PC with different ratios of palm oil. Liposomal formulations containing 5% and 10% of palm oil were made through the freeze-thaw method, and then the TEM images revealed satisfactory morphology and formation of LUVs, respectively. Liposomal size distribution, zeta potential, and stability remain in the acceptable range. The HPLC results confirm the optimal drug loading through

pH gradient technique and sophisticated in vitro release profile as well.

The finest cellular uptake was observed on MCF-7 and MDA-MBA 231 cell lines through 24 hr furthermore, cytotoxicity assay confirms the more effectiveness of the liposomal Trilaciclib containing palm oil.

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