



## Silver Diamine Fluoride---A Double Edgedsword

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**ABSTRACT:** Metallic silver was used in past for disinfecting & preserving drinking water. In metallic state silver is inert. In aqueous media silver is ionised. In its ionic state silver enters bacterial cells by endocytosis and causes membrane inactivation. Silver diamine fluoride [Ag (NH<sub>3</sub>)<sub>2</sub>F] has two pronged effect in the shape of antimicrobial effect of silver while caries arrest by fluoride. SDF is effective in carious lesion in primary teeth following the standards set by WHO mellenium goals. SDF causes mineralization of peritubular dentin and increases dentin microhardness. Due to ease of manipulation, controlling pain and infection, non-invasive process etc SDF is termed as silver bullet. SDF inactivates metalloproteinases responsible for collagen breakdown in caries process. 38% SDF solution is effective in inactivating MMP2, MMP8 & MMP9.

**Key words:** SDF; silver ion; endocytosis; dentin microhardness; metalloproteinases; MMP2; MMP8; MMP9.

### I. INTRODUCTION

Silver has been used in human healthcare and medicine since long. But, it is not used for any nutritional benefit. The physiological level is (< 2.3 µg/l) of silver, higher in individual subjected to silver for long periods [1,2]. Water was preserved and disinfected for drinking using metallic silver. Silver pots were used for storing and drinking water by Alexander the Great (335 BC) when he used to go on his many campaigns [3,4]. Silver was used to preserve and disinfect water aboard the Apollo spacecraft [5], the MIR space station [9] and the NASA space shuttle [6-8]. In the 1800s silver was used as an antiseptic for post surgical infections, in dentistry, wound therapy and medical devices.

Later penicillin, sulfonamide and mafenide antibiotics replaced silver and its compounds in most parts of the world for 40 years after the Second World War [10-12]. The

emergence of resistant strain organisms such as Pseudomonas aeruginosa and methicillin resistant Staphylococcus aureus (MRSA) to penicillin and sulfonamide drugs led to the need for new antibiotics. Moyer et al., is credited for introducing the use of silver nitrate in the 1960s [13]. Following this, silver sulfadiazine (SSD) in 1968 [14], was introduced, as efficacious for local application. SSD (Figure 1) has a broad spectrum of antimicrobial action. Pure metallic silver is inert and reacts with human tissue or kill microorganisms only after ionisation. Metallic silver ionises into silver ions in aqueous media and is bioactive in the form of Ag<sup>+</sup>

ion. Recent developments have shown that the presence of higher halide concentration with fewer Ag<sup>+</sup> ions results in the formation of anionic silver complexes (AgX<sub>2</sub><sup>-</sup>, where X = Cl<sup>-</sup>), which are soluble in aqueous media and are bio-active [15]. Maratech Holdings reports in vivo experimental evidence for the antiviral properties of Ag<sub>4</sub>O<sub>4</sub> [16,17]. The mechanism of action of antimicrobial effects of oxidation states of silver needs further investigation.

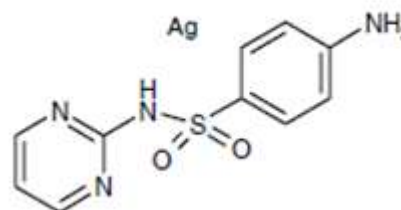


Figure 1 silver sulphadiazine.

In its metallic form, silver is inert and exhibits no biocidal action. But metallic silver ionizes in aqueous media or tissue fluids to release Ag or other biologically active ions. The ionised silver possess a strong affinity for sulphhydryl groups and protein residues on cell membranes. Silver at a very low concentration (1 ppm) possess



antimicrobial property and is important for the widely known oligodynamic effect coined by von Naegeli in 1895.

Mode of antimicrobial action of silver in sensitive organisms are complex and not well defined (fig. 2). Attachment of silver to cell membranes and endocytosis is an obligatory first step; silver attaches to negatively charged receptors, notably disulphide, amino, imidazole, carbonyl and phosphate residues on cell membranes resulting in intracellular absorption by endocytic vacuoles and phagocytosis. Intracellular absorption of silver paralyzes membrane-related enzymes like phosphomannose isomerase which in turn modifies the bacterial cell envelope and its efficacy of controlling the inward diffusion of nutrients (e.g. phosphates, succinates) and interferes with outward passage of essential electrolytes and metabolites.

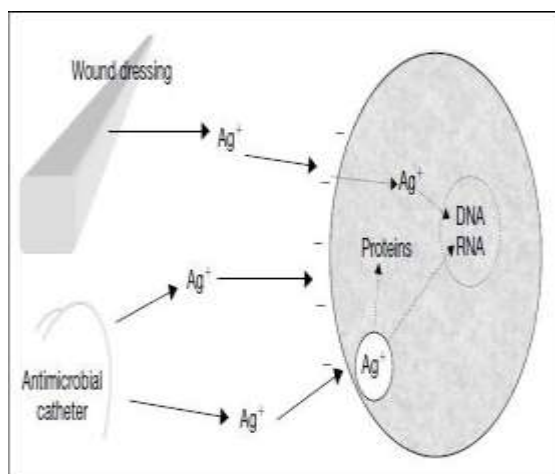


Figure 2 showing  $Ag^+$  (silver ion), invading bacterial cell membrane by endocytosis.

Membrane damage has been identified by pitting and increased permeability as a prelude to lethality. Principal intracellular effect of silver ion is possibly the ability to paralyze key intracellular enzyme systems. The trace metals and electrolytes are inactivated causing defective respiratory pathways and RNA and DNA replication [18].

$AgNO_3$  is used medically in eye drops to prevent infections in newborns, and in dentistry it is often used in stomatological treatments for mouth ulcers [19,20]. In 1969, silver diamine fluoride [ $F(NH_3)_2Ag$ ] (SDF) solution was synthesized for dental treatments [21-23]. Since then, it has been used in Japan as Saforide R Solution for application to carious lesions due to its capacity as an antimicrobial agent and to stabilize caries processes, particularly in primary teeth, thanks to which it has an important role in

pediatric dentistry [24,25]. SDF is a colorless solution which is used at 38-40%, pH 8-10. On contacting the caries surface it produces calcium fluoride ( $CaF_2$ ) and silver phosphate ( $Ag_3PO_4$ ) [25]. The F:Ag ion ratio is 44,800:255,000 ppm [24,26,27].

## II. CARIES PREVENTION IN PRIMARY TEETH

SDF has been observed as biocompatible and competent caries controlling compound as per the standards laid down by the WHO Millennium Goals and the US Institute [28]. Arresting progression of carious lesion and occurrence of new lesions in primary teeth by SDF has been shown in clinical trials [29,30] and root caries in permanent teeth [31]. SDF is also capable of enhancing microhardness [32-34] and maturation of dentine shown in many laboratory studies [33,35,36].

## III. ACT FOR DISADVANTAGED COMMUNITY

Untreated dental caries is a global pandemic [37]. Dental caries remains untreated due to financial constraint, inadequate availability of oral care & expensive restorative treatment; this in turn is affecting average health, social life, and learning abilities of children of low income states [38]. Arresting Caries Treatment (ACT) has been adopted for combating unattended carious lesions in under privileged children [39]. Silver diamine fluoride (SDF),  $Ag(NH_3)_2F$ , has been in use to halt caries as long as before 1969 [40,41]. 38% SDF (44,800 ppm F) applied once a year to carious deciduous anterior teeth of Chinese preschool children have been found to be significantly more successful in stopping progression of caries and preventing new caries than single applications of sodium fluoride varnish (22,600 ppm F) [42]. 38% SDF application biyearly proved effective in stopping lesion progression and new lesion in teeth of adolescent age groups of Cuban children over a three-year period [43].

Importance of silver diamine fluoride centers around its 5 presumed effects (Bedi and Sardo-Infirri, 1999) [44]: 1. control of pain and infection, 2. ease and simplicity of use (paint on), 3. affordability of material (pennies per application), 4. minimal requirement for personnel time and training (one minute, once per year), and the fact that 5. it is non-invasive. In this sense, SDF has the inherent specific efficacy to be a "silver-fluoride bullet," simultaneously halting the cariogenic process and preventing caries.



➤The caries preventive effects of SDF are emphasized in terms of the World Health Organization (WHO) Millennium Development 1. Goals for Health (Wagstaff and Claeson, 2004),<sup>45</sup> and in particular the ➤2. oral health goals (Hobdell et al., 2003)<sup>46</sup>. These aims can be reached by satisfying provision of a basic oral health package, consisting of: 1. emergency care, ➤2. prevention, and ➤. Cost-effective interventions, in that order (Frencken et al., 2008)<sup>47</sup>

Silver diamine fluoride (SDF) was used in clinical trials to halt dentin caries and the results were prospective [48-50] in Children, specially fearful children of tender age, unable to cooperate for long and not very simple restorative procedures. There are various methods proposed to arrest dental caries in children, such as xylitol gum chewing and the use of fluoridated agents.

#### IV. ADVANTAGES OF SDF

As silver diamine fluoride is cost-effective and simple procedure it is given much importance in child dental patients. Reduced pain and infection, ease of manipulation, minimum cost, non-invasive technique and less time and training are the benefits of caries control with SDF. According

to a systematic review the objective of millennium by WHO and United States Institute of Medicine's standard of 21<sup>st</sup> century are satisfied by SDF [51].

#### V. REVIEW OF LITERATURE

According to studies infected dentin and affected dentin are distinct layers of carious dentin differ chemically as well as structurally. [52,53]. The superficial infected layer harbouring bacteria while the inner affected dentine layer is softened by plaque acid but relatively less infected. Due to higher content of mineral salts this affected layer can be re-mineralised in favourable conditions [54].

Intertubular dentin hypercalcification and tubular occlusion results following treatment with SDF (Fig. 3). Microscopically sealing of tubules following SDF application results in reduced permeability (Fig. 4). Dentin under the SDF treated lesion appears normal while pulp shows inflammatory infiltrate and tertiary dentin formation (Fig. 5). Mei et al [55] showed silver phosphate formation and precipitation with SDF in patients with increased caries. Reduction in mineral loss from carious lesion occurs due to formation of protective layer of calcium fluoride, silver phosphate and silver with less soluble protein [55,56].

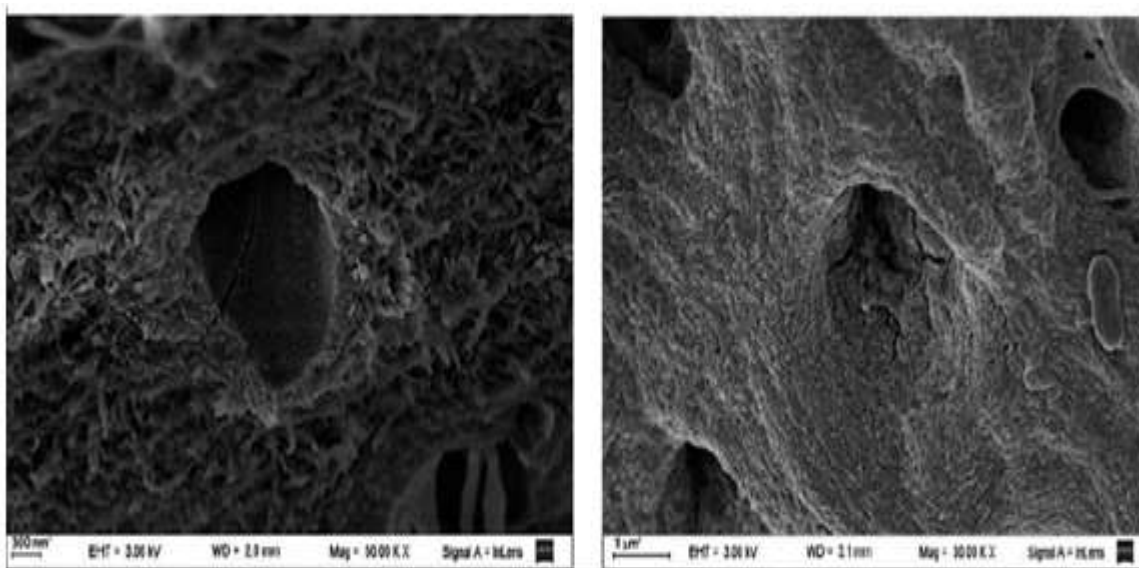


Figure 3: showing dentin structures before SDF application (left), and partially occluded tubules and hypermineralized inter-tubular dentin following SDF application in SEM photograph (right)

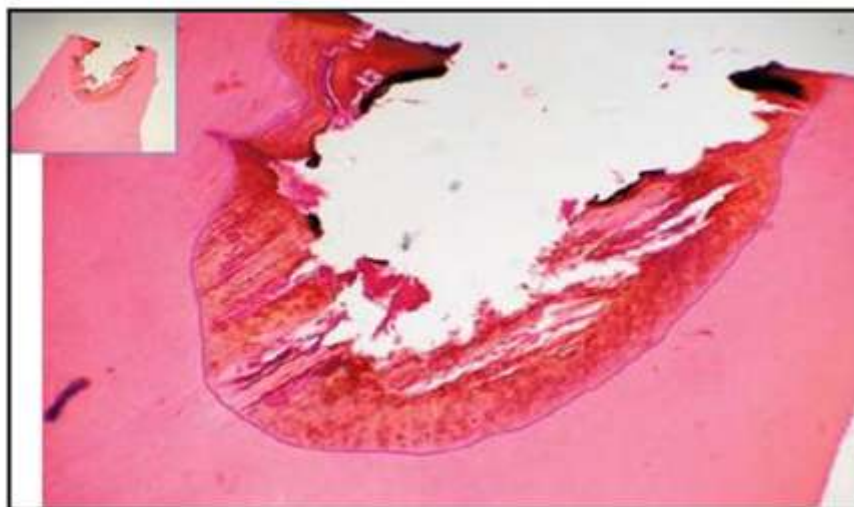


Figure 4: microscopical image of dental tissue following demineralization and staining, image showing deposits of silver in carious dentin, line of demarcation limits SDF – 100X enlargement.

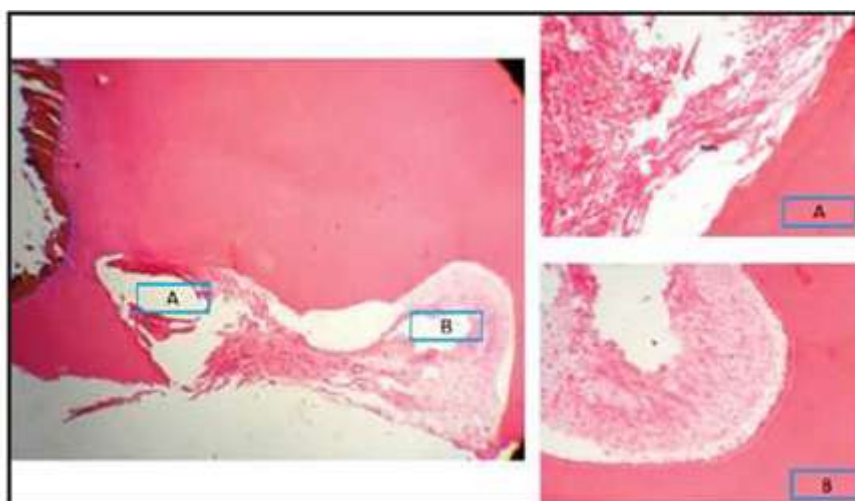


Figure 5: image showing histopathological tissue, following SDF application after demineralization and staining, affected pulp tissue in A (on left), on right higher resolution showing pulp inflammatory cells and fibroblasts, B is pulp with unaffected dentin, Pulpal circulation seen in higher resolution

38% SDF causes decrease in mineral loss from carious lesion as well as collagen destruction and thus stops lesion progression, Mei et al [55]. Besides, high concentration of silver and fluoride prevent growth of cariogenic bacteria in carious plaque---thus SDF acts as double-edged weapon against dental caries. SDF blocks metalloproteinases and prevents disintegration of collagen in caries, and thus protects against dentin disintegration. In vitro study by Mei et al [55] shows that SDF primarily interacts with hydroxyapatite of dentin and the reaction product calcium fluoride prevents against caries. As microhardness depends on mineral content of dentin, changes due to SDF application may thus also be determined by microhardness.

## VI. POSSIBLE MECHANISM OF ACTION OF SDF IN CARIES INHIBITION

Caries progression is a factor of decalcification of hydroxyapatite and destruction of collagen matrix. collagenases, microbial enzymes cause destruction of organic matrix. Critical role of metalloproteinases in dentin disintegration has been found in different contemporary studies [57]. Matrix metalloproteinase's or matrixins are calcium-dependent zinc containing endopeptidases [58]. Characteristically MMP is made up of prodomain, prodomain, hinge, catalytic domain and hemopexin



domain. An eighty amino acid compound prodomain is attached to prodomain. MMPs are inactive zymogens, have cysteine lock present in prodomain prevents their intracellular activation [59]. MMPs can be activated by proteinases [57], chemical agents and, in caries state by acidic pH of the environment. Prodomain is connected with the catalytic domain by a hinge and the catalytic domain holds an active Zn-cysteine binding site. The activation of MMP occurs by breaking Zn cysteine interaction [58]. One fibronectin domain present in MMP-2 (gelatinase A) and MMP-9 (gelatinase B), has a strong affinity for gelatin. The catalytic domain is connected to the hemopexin domain. The hemopexin domain of MMP-2 and MMP-9 activates the tissue inhibitor enzymes for metalloproteinases [59]. Disintegration of extracellular collagen matrix is mediated by MMPs in presence of zinc ion ( $Zn^{2+}$ ) which acts as a cofactor [57]. In health MMPs are present in dentin matrix [60, 61] or in saliva [62]. Action of MMPs is stimulated in low pH environment namely lactate release by causative bacteria [57]. MMP-8 (neutrophil collagenase) is capable of degrading triple helical fibrillar collagens into distinctive  $\frac{3}{4}$  and  $\frac{1}{4}$  fragments. MMP-2 and MMP-9 are gelatinase, which cause breakdown of type IV collagen. Collagenolysis is an integral part of dentin caries during which stimulation of MMP-2, MMP-8 and MMP-9 has been shown to have a crucial role [57]. So inactivation of MMPs is an important mechanism of caries arrest.

## VII. CONCLUSION:

38% SDF has highest efficacy in inactivating effect on MMP2, MMP8 and MMP9. Comparative to silver nitrate ( $AgNO_3$ ) and sodium fluoride (NaF) SDF possess greater inhibition on MMPs. Thus success of 38% of SDF in causing caries arrest is attributed to its inhibition on MMPs in clinical trials.

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